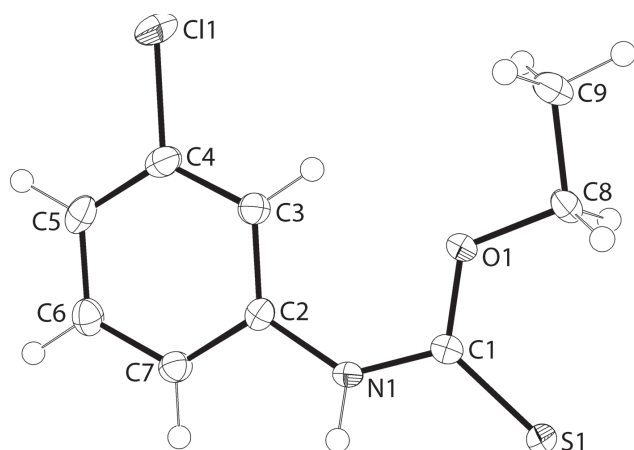


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Crystal structure of *N*-(3-chlorophenyl)ethoxy-carbothioamide, C₉H₁₀ClNOS



<https://doi.org/10.1515/ncrs-2017-0403>

Received December 12, 2017; accepted March 18, 2018; available online March 30, 2018

Abstract

C₉H₁₀ClNOS, monoclinic, *C*2/*c* (no. 15), *a* = 18.8783(6) Å, *b* = 4.5518(2) Å, *c* = 23.6739(8) Å, β = 91.501(4)°, *V* = 2033.61(13) Å³, *Z* = 8, *R*_{gt}(*F*) = 0.033, *wR*_{ref}(*F*²) = 0.082, *T* = 100(2) K.

CCDC no.: 1830621

The asymmetric unit of the title crystal structure is shown in the figure. Tables 1 and 2 contain details of the measurement method and a list of the atoms including atomic coordinates and displacement parameters.

Source of materials

3-Chlorophenyl isothiocyanate (Sigma Aldrich; 2.5 mmol, 0.33 mL) was added to NaOH (Merck; 2.5 mmol, 0.10 g) in EtOH (Merck; 3 mL) and the mixture was stirred at room temperature for 2 h, followed by the addition of excess 5 M HCl solution. The resulting mixture was stirred for another 1.5 h. The final product was extracted with chloroform (Merck;

Table 1: Data collection and handling.

Crystal:	Colourless prism
Size:	0.30 × 0.22 × 0.05 mm
Wavelength:	Mo Kα radiation (0.71073 Å)
μ:	5.4 cm ⁻¹
Diffractometer, scan mode:	Bruker SMART APEX, ω scans
2θ _{max} , completeness:	55°, >99%
<i>N</i> (<i>hkl</i>) _{measured} , <i>N</i> (<i>hkl</i>) _{unique} , <i>R</i> _{int} :	9015, 2327, 0.035
Criterion for <i>I</i> _{obs} , <i>N</i> (<i>hkl</i>) _{gt} :	<i>I</i> _{obs} > 2 σ(<i>I</i> _{obs}), 2015
<i>N</i> (<i>param</i>) _{refined} :	122
Programs:	Bruker programs [1, 2], SHELX [3, 4], ORTEP [5]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{iso} */ <i>U</i> _{eq}
Cl1	0.39796(3)	0.55128(12)	0.24380(2)	0.03764(15)
S1	0.36135(2)	-0.36422(9)	-0.00973(2)	0.01921(12)
O1	0.42262(6)	-0.0977(2)	0.07864(5)	0.0194(2)
N1	0.30855(7)	0.0069(3)	0.06370(6)	0.0163(3)
H1N	0.2704(7)	-0.028(4)	0.0438(7)	0.020*
C1	0.36499(8)	-0.1440(3)	0.04656(6)	0.0157(3)
C2	0.30022(8)	0.2121(3)	0.10787(6)	0.0163(3)
C3	0.35058(8)	0.2705(4)	0.15030(7)	0.0199(3)
H3	0.3948	0.1711	0.1513	0.024*
C4	0.33465(9)	0.4769(4)	0.19103(7)	0.0218(4)
C5	0.27084(9)	0.6245(4)	0.19176(7)	0.0220(3)
H5	0.2614	0.7648	0.2203	0.026*
C6	0.22107(9)	0.5617(4)	0.14962(7)	0.0223(4)
H6	0.1765	0.6586	0.1494	0.027*
C7	0.23542(8)	0.3593(3)	0.10777(7)	0.0200(3)
H7	0.2009	0.3206	0.0788	0.024*
C8	0.48604(8)	-0.2722(4)	0.06945(7)	0.0223(4)
H2A	0.5083	-0.2145	0.0337	0.027*
H2B	0.4742	-0.4839	0.0676	0.027*
C9	0.53472(9)	-0.2092(5)	0.11901(8)	0.0339(5)
H3A	0.5780	-0.3261	0.1159	0.051*
H3B	0.5111	-0.2605	0.1540	0.051*

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15 mL) and left for evaporation at room temperature, yielding colourless crystals after 2 weeks. M.p. (Krüss KSP1N): 349–350 K. IR (Perkin Elmer Spectrum 400 F T Mid-IR/Far-IR; cm⁻¹): 3248 (s) ν(N–H), 1476 (s) ν(C–N), 1199 (s) ν(C=S), 1040 (s) ν(C–O). Elem. Anal. (Perkin Elmer PE 2400 CHN):

Calc. for C₉H₁₀ClNOS: C, 50.11; H, 4.67; N, 6.49%. Found: C, 50.21; H, 4.55; N, 6.69%.

Experimental details

The C-bound H atoms were geometrically placed (C–H = 0.95–0.99 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5 U_{\text{eq}}(\text{C})$. The N-bound H-atoms were located in a difference Fourier map but were refined with a distance restraint of N–H = 0.88 ± 0.01 Å, and with $U_{\text{iso}}(\text{H})$ set to $1.2 U_{\text{eq}}(\text{N})$. Owing to poor agreement, two reflections, i.e. (2 0 0) and (0 0 2), were omitted from the final cycles of refinement, presumably owing to interference from the beam-stop.

Comment

The structural landscape of molecules of the general formula ROC(=S)N(H)R', for R, R' = alkyl/aryl, has been summarized recently [6]. The overwhelming majority of structures, including a more recently reported derivative [7], share common features such as a *syn*-disposition of the thioamide-N–H and thione-S atoms [6, 7] with the exceptions, i.e. with anti-dispositions of these atoms, arising owing to the dictates of thioamide-N–H···O, N hydrogen bonding, as the O– and/or N-bound substituents carry good oxygen or nitrogen hydrogen bond acceptors [6]. In continuation of these studies, the title compound EtOC(=S)N(H)(3-ClPh) was investigated by X-ray crystallography.

The molecular structure is shown in the Figure (50% displacement ellipsoids) and this features the anticipated *syn*-disposition of the thioamide-N–H and thione-S atoms. The central CNOS chromophore is planar and forms a dihedral angle of $7.94(9)^\circ$ with the N-bound 3-chlorophenyl ring. The whole molecule is practically planar with the r.m.s. for the 13 fitted non-hydrogen atoms being 0.0594 Å with maximum deviations of $0.1186(12)$ and $0.0973(12)$ Å for the O1 and N1 atoms, respectively. Overall, the molecule has a U-shape as

the ethoxy and phenyl-chloro groups lie to the same side of the molecule.

In the molecular packing, the *syn*-disposition of the thioamide-N–H and thione-S atoms facilitates the formation of thioamide-N–H···S(thione) hydrogen bonds between centrosymmetrically related molecules and therefore, eight-membered thioamide synthons $\{\cdots\text{HNCS}\}_2$; N1–H1n···S1ⁱ: $2.641(14)$ Å and $164.7(14)^\circ$ for symmetry operation *i*: $1/2 - x, -1/2 - y, -z$. Globally, in the crystal, dimeric aggregates stack in columns parallel to the *b* axis.

Acknowledgements: The University of Malaya's X-ray laboratory is thanked for the data collection. Sunway University is thanked for support of biological and crystal engineering studies of metal alkoxycarbothioamides.

References

1. Krause, L.; Herbst-Irmer, R.; Sheldrick, G. M.; Stalke, D.: Comparison of silver and molybdenum microfocus X-ray sources for single-crystal structure determination. *J. Appl. Cryst.* **48** (2015) 3–10.
2. Bruker. APEX2 and SAINT. Bruker AXS Inc., Madison, WI, USA (2008).
3. Sheldrick, G. M.: A short history of SHELX. *Acta Crystallogr.* **A71** (2015) 3–8.
4. Sheldrick, G. M.: Crystal structure refinement with SHELXL. *Acta Crystallogr.* **C71** (2015) 3–8.
5. Farrugia, L. J.: WinGX and ORTEP for Windows: an update. *J. Appl. Cryst.* **45** (2012) 849–854.
6. Jotani, M. M.; Yeo, C. I.; Tiekink, E. R. T.: A new monoclinic polymorph of *N*-(3-methylphenyl) ethoxycarbothioamide: crystal structure and Hirshfeld surface analysis. *Acta Crystallogr.* **E73** (2017) 1889–1897.
7. Yeo, C. I.; Tiekink, E. R. T.: Crystal structure of *N*-(2-methylphenyl)ethoxycarbothioamide, C₁₀H₁₃NOS. *Z. Kristallogr. – NCS* **233** (2018) 299–301.