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Crystal structure of the co-crystal O-isopropyl phenylcarbamothioate -4,4'-bipyridine (2/1), C₁₅H₁₇N₂OS



https://doi.org/10.1515/ncrs-2017-0374 Received November 28, 2017; accepted March 18, 2018; available online April 5, 2018

Abstract

a = 5.8595(3) Å, C₁₅H₁₇N₂OS, triclinic, ΡĪ (no. 2), b = 9.2967(4) Å, c = 14.0545(8) Å, $\alpha = 83.320(4),$ $\beta =$ $V = 732.71(7) \text{ Å}^3$, 89.240(4), $\gamma = 74.524(4),$ Z = 2, $R_{\rm gt}(F) = 0.0458, wR_{\rm ref}(F^2) = 0.1231, T = 100$ K.

CCDC no.: 1830651

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Table 1: Data collection and handling.

Crystal:	Block, colourless
Size:	$0.10\times0.05\times0.05~\text{mm}$
Wavelength:	Mo <i>Kα</i> radiation (0.71073 Å)
μ:	0.22 mm^{-1}
Diffractometer, scan mode:	SuperNova, ω -scans
θ_{\max} , completeness:	27.6°, >99%
N(hkl) _{measured} , N(hkl) _{unique} , R _{int} :	5109, 3354, 0.035
Criterion for I _{obs} , N(hkl) _{gt} :	$I_{\rm obs} > 2 \ \sigma(I_{\rm obs})$, 2762
N(param) _{refined} :	178
Programs:	CrysAlis ^{PRO} [1], SHELX [2, 3],
	WinGX and ORTEP [4]

The crystal structure is shown in the figure. Tables 1 and 2 contain details on crystal structure and measurement conditions and a list of the atoms including atomic coordinates and displacement parameters.

Source of material

4,4'-Bipyridine (Merck; bpy; 0.05 g, 0.32 mmol) and 2.1 mole equivalent of i-PrOC(=S)N(H)Ph [5] (0.11 g, 0.67 mmol), each in dichloromethane (10 mL), were mixed and stirred at 323 K for 3 h. The resulting mixture was left for evaporation at room temperature after which colourless blocks were deposited. Anal. Calc. for C₁₅H₁₇N₂OS: C, 65.90; H, 6.27; N, 10.25. Found: C, 63.28; H, 6.66; N, 10.24. IR (cm⁻¹): 3164 (br) ν(N-H); 1593 (s) v(C=C; byp); 1448 (s) v(C-N); 1393 (s) v(C-N, bpy); 1204 (s) ν (C=S); 1087 (vs) ν (C-O).

Experimental details

The C-bound H atoms were geometrically placed (C-H = 0.95 - 1.00 Å) and refined as riding with $U_{iso}(H) = 1.2 - 1.00$ $1.5U_{eq}(C)$. The N-bound H-atom was located in difference Fourier map but refined with a distance restraint of $N-H = 0.88 \pm 0.01$ Å, and with $U_{iso}(H)$ set to $1.2U_{equiv}(N)$.

Discussion

The alkoxycarbothioamides, i.e. molecules of the general formula ROC(=S)N(H)R' for R, R' = alkyl/aryl [6], find interest in crystal engineering endeavours in terms of their ability to form thioamide-N-H···S(thione) hydrogen bonds and the

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	x	у	z	U _{iso} */U _{eq}
S 1	0.82329(9)	-0.06274(5)	0.19555(4)	0.02853(15)
01	0.5643(2)	0.20073(14)	0.10848(10)	0.0257(3)
N1	0.4505(3)	0.13231(16)	0.25485(11)	0.0196(3)
H1N	0.456(4)	0.065(3)	0.3029(17)	0.028(6)*
C2	0.2515(3)	0.25973(18)	0.25604(12)	0.0167(3)
С3	0.0502(3)	0.23730(19)	0.30264(13)	0.0202(4)
H3	0.0485	0.1394	0.3304	0.024*
C4	-0.1482(3)	0.3574(2)	0.30864(13)	0.0232(4)
H4	-0.2847	0.3413	0.3406	0.028*
C5	-0.1476(3)	0.5007(2)	0.26827(13)	0.0244(4)
H5	-0.2834	0.5828	0.2720	0.029*
C6	0.0533(3)	0.5226(2)	0.22247(14)	0.0241(4)
H6	0.0539	0.6207	0.1947	0.029*
C7	0.2542(3)	0.40392(18)	0.21635(13)	0.0197(4)
H7	0.3915	0.4209	0.1855	0.024*
C1	0.6066(3)	0.09765(18)	0.18471(13)	0.0195(4)
C8	0.7202(3)	0.1838(2)	0.02449(14)	0.0279(4)
H8	0.7873	0.0747	0.0178	0.034*
C9	0.5605(4)	0.2597(3)	-0.06113(16)	0.0412(6)
H9A	0.4302	0.2126	-0.0643	0.062*
H9B	0.6521	0.2489	-0.1200	0.062*
H9C	0.4959	0.3666	-0.0544	0.062*
C10	0.9150(4)	0.2537(3)	0.03901(17)	0.0477(6)
H10A	1.0055	0.2036	0.0973	0.072*
H10B	0.8484	0.3605	0.0455	0.072*
H10C	1.0198	0.2432	-0.0161	0.072*
N2	0.4473(3)	-0.11967(17)	0.40417(12)	0.0265(4)
C11	0.6235(3)	-0.1965(2)	0.46444(15)	0.0271(4)
H11	0.7364	-0.1469	0.4818	0.033*
C12	0.6514(3)	-0.34441(19)	0.50318(14)	0.0230(4)
H12	0.7810	-0.3935	0.5455	0.028*
C13	0.4892(3)	-0.42063(17)	0.47995(12)	0.0171(3)
C14	0.3044(4)	-0.3405(2)	0.41794(16)	0.0315(5)
H14	0.1872	-0.3864	0.4001	0.038*
C15	0.2921(4)	-0.1928(2)	0.38220(16)	0.0346(5)
H15	0.1649	-0.1408	0.3394	0.042*

influence of this upon molecular conformation [5–7] as well as their propensity to form co-crystals [8]. As a continuation of the latter studies, the title 2:1 co-crystal was isolated.

As indicated from the figure (50% displacement ellipsoids), there is a thioamide-N–H····S(thione) hydrogen bond connecting the co-crystal coformers [N1–H···S1 = 2.11(3) Å and 170(2)°]. The strictly planar 4,4′-bipyridyl molecule is situated about a centre of inversion with the unlabelled atoms of the figure related by the symmetry operation: 1 - x, -1 - y, 1 - z. The three-molecule aggregate resembles that observed in the previously reported 2:1 co-crystal of the title carbothioamide with trans-1,2-bis(4-pyridyl)ethene [8]. Indeed, all known co-crystals of this type,

i.e. $[ROC(=S)N(H)R']_2$ (bipyridyl-type molecule), six in all, feature a similarly assembled three-molecule aggregate [8]. The *svn* disposition of the thioamide-N–H and thione-S atoms is consistent with that observed in the three other known structures containing the identical i-PrOC(=S)N(H)Ph molecule. An overlap diagram (not shown) of the molecule reported herein, the parent compound [5], the molecule in the 2:1 cocrystal with trans-1,2-bis(4-pyridyl)ethene [8] and that with the thione-S coordinating a (Ph₃P)₂CuCl entity [9] shows a remarkable consistency of conformation. The major difference arises, in fact, for the *i*-Pr group in the title compound which is rotated about the O–C axis by approximately 60° compared to the three literature molecules. The dihedral angles between the central CNOS residue and appended N-bound phenyl ring vary from apprimxately 28° in the copper(I) complex [9] to 47° in the co-crystal [8].

Acknowledgements: The University of Malaya's X-ray laboratory is thanked for the data collection. Sunway University is thanked for support of biological and crystal engineering studies of metal thiocarbamides.

References

- 1. Agilent Technologies. CrysAlis PRO. Agilent Technologies, Santa Clara, CA, USA (2010)
- Sheldrick, G. M.: A short history of SHELX. Acta Crystallogr. A64 (2008) 112–122.
- Sheldrick, G. M.: Crystal structure refinement with SHELXL. Acta Crystallogr. C71 (2015) 3–8.
- Farrugia, L. J.: WinGX and ORTEP for Windows: an update. J. Appl. Cryst. 45 (2012) 849–854.
- Kuan, F. S.; Mohr, F.; Tadbuppa, P. P.; Tiekink, E. R. T.: Principles of crystal packing in *O*-isopropyl-*N*aryl-thiocarbamides: iPrOC(=S)N(H)C₆H₄-4-Y: Y = H, Cl, and Me. CrystEngComm 9 (2007) 574–581.
- Jotani, M. M.; Yeo, C. I.; Tiekink, E. R. T.: A new monoclinic polymorph of *N*-(3-methylphenyl) ethoxycarbothioamide: crystal structure and Hirshfeld surface analysis. Acta Crystallogr. **E73** (2017) 1889–1897.
- Ho, S. Y.; Bettens, R. P. A.; Dakternieks, D.; Duthie, A.; Tiekink, E. R. T.: Prevalence of the thioamide {···H-N-C=S}₂ synthon – solid-state (X-ray crystallography), solution (NMR) and gas-phase (theoretical) structures of *O*-methyl-*N*-arylthiocarbamides. CrystEngComm 7 (2005) 682–689.
- Ellis, C. A.; Miller, M. A.; Spencer, J.; Zukerman-Schpector, J.; Tiekink, E. R. T.: Co-crystallization experiments of thiocarbamides with bipyridine-type molecules. CrystEngComm 11 (2009) 1352–1361.
- Yeo, C. I.; Halim, S. N. A.; Ng, S. W.; Tan, S. L.; Zukerman-Schpector, J.; Ferreira, M. A. B.; Tiekink, E. R. T.: Investigations of putative arene-C—H···π(quasi-chelate ring) interactions in copper(I) crystal structures. Chem. Commun. **50** (2014) 5984–5986.