

CRYSTAL TRANSFORMATION AND POLYMORPHS OF BIS(N-HYDROXYETHYL-N-ISOPROPYL DITHiocarbamato)CADMIUM(II)

TAN YEE SENG

RESEARCH CENTRE FOR CRYSTALLINE MATERIALS
SUNWAY UNIVERSITY

INTRODUCTION

- Crystal Transformation
- Polymorph
- Cadmium Dithiocarbamate

CRYSTAL TRANSFORMATION

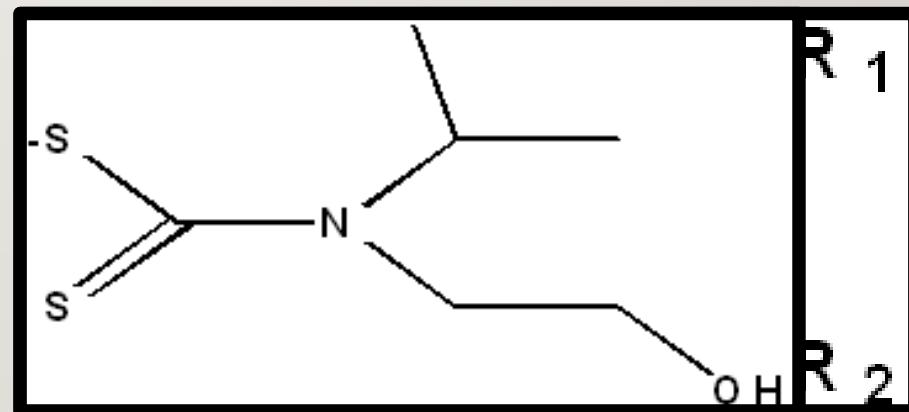
- One form to another
- Factor
 - Temperature
 - Solvent
 - Pressure
 - pH
 - etc.

POLYMORPHISM

- Different molecular arrangement
 - Molecule concerned
 - Solvent/guest
- How to identify
 - Unit Cell
 - PXRD pattern

CADMIUM DITHiocarbamate

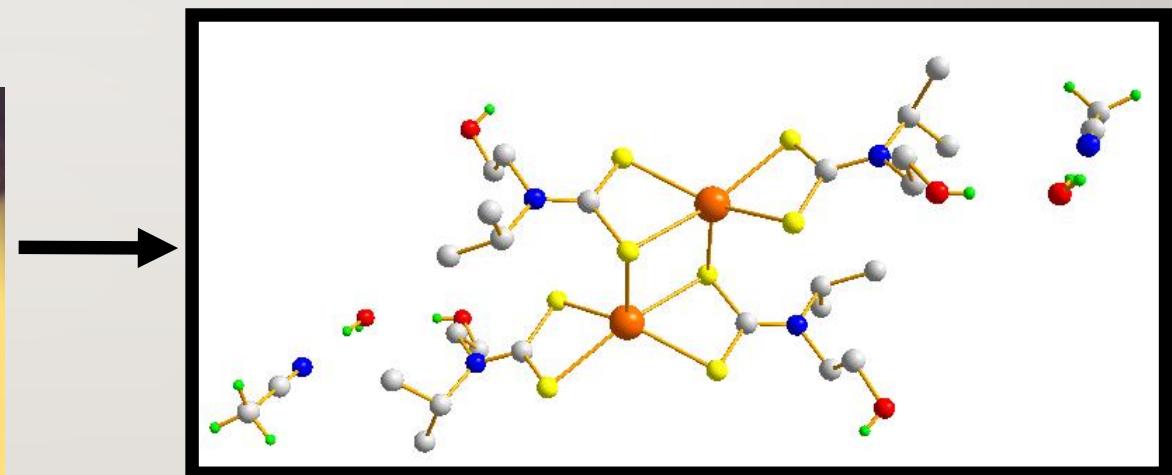
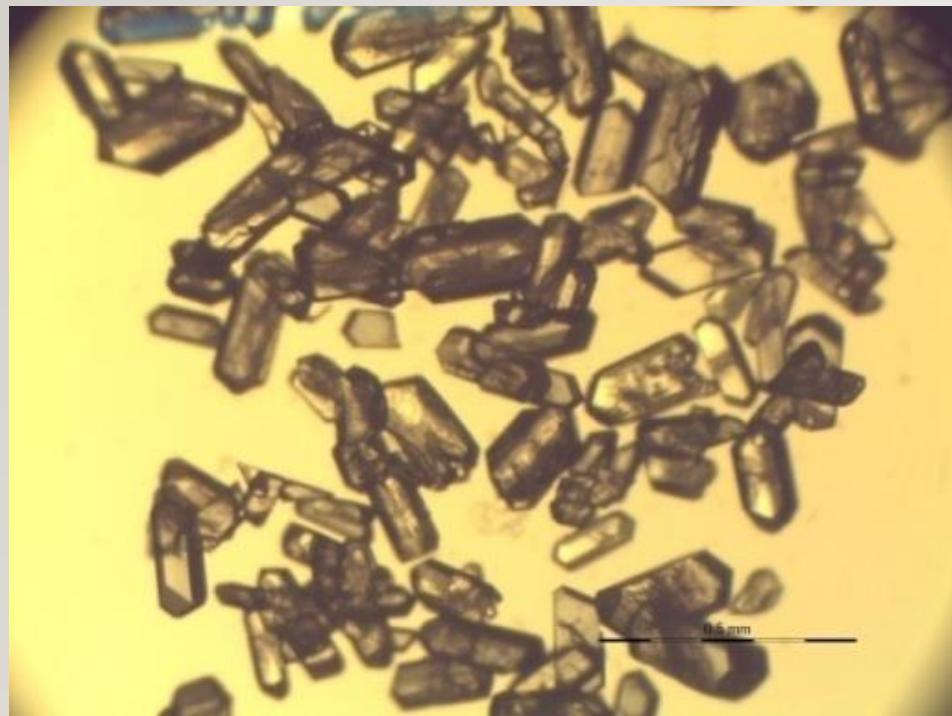
- Cadmium Metal Ion (Cd^{2+})
- Dithiocarbamate Ligand



OBJECTIVE

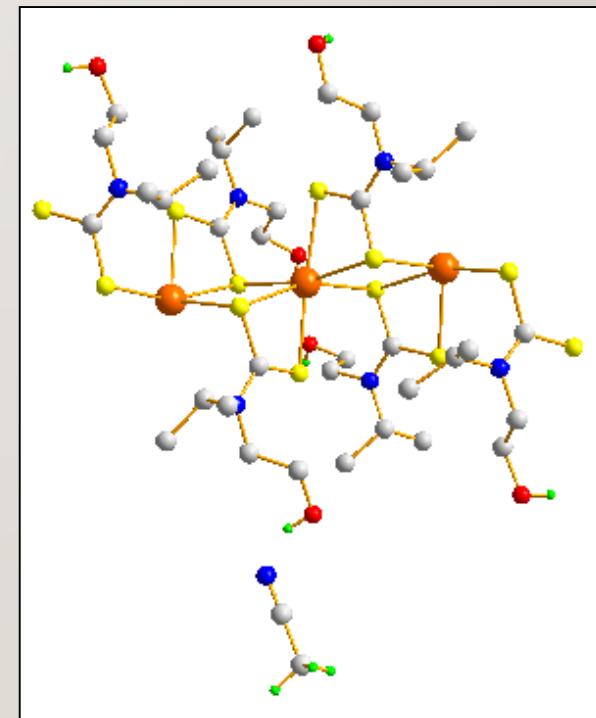
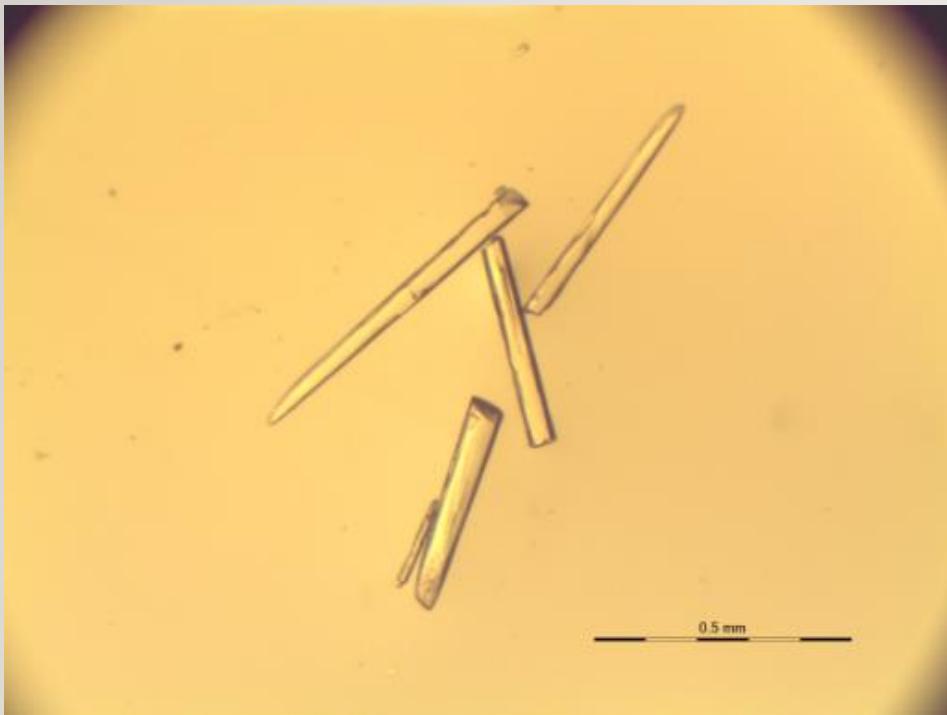
- Intermolecular interaction
- Solvent free

SLOW EVAPORATION (ACETONITRILE)



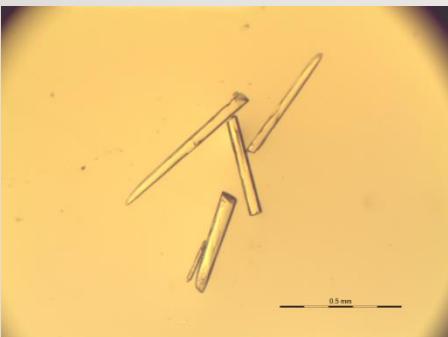
Characterisation and analysis?

SLOW EVAPORATION (ACETONITRILE)



CRYSTAL TRANSFORMATION 1

Day1



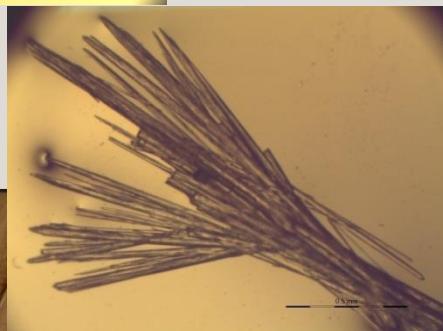
α

Day2

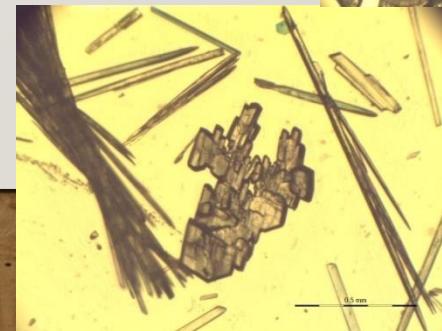


Day5

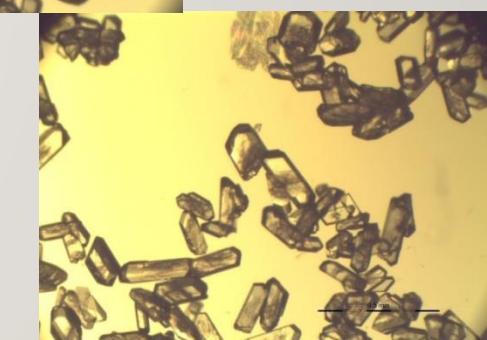
Day3



γ



Day4



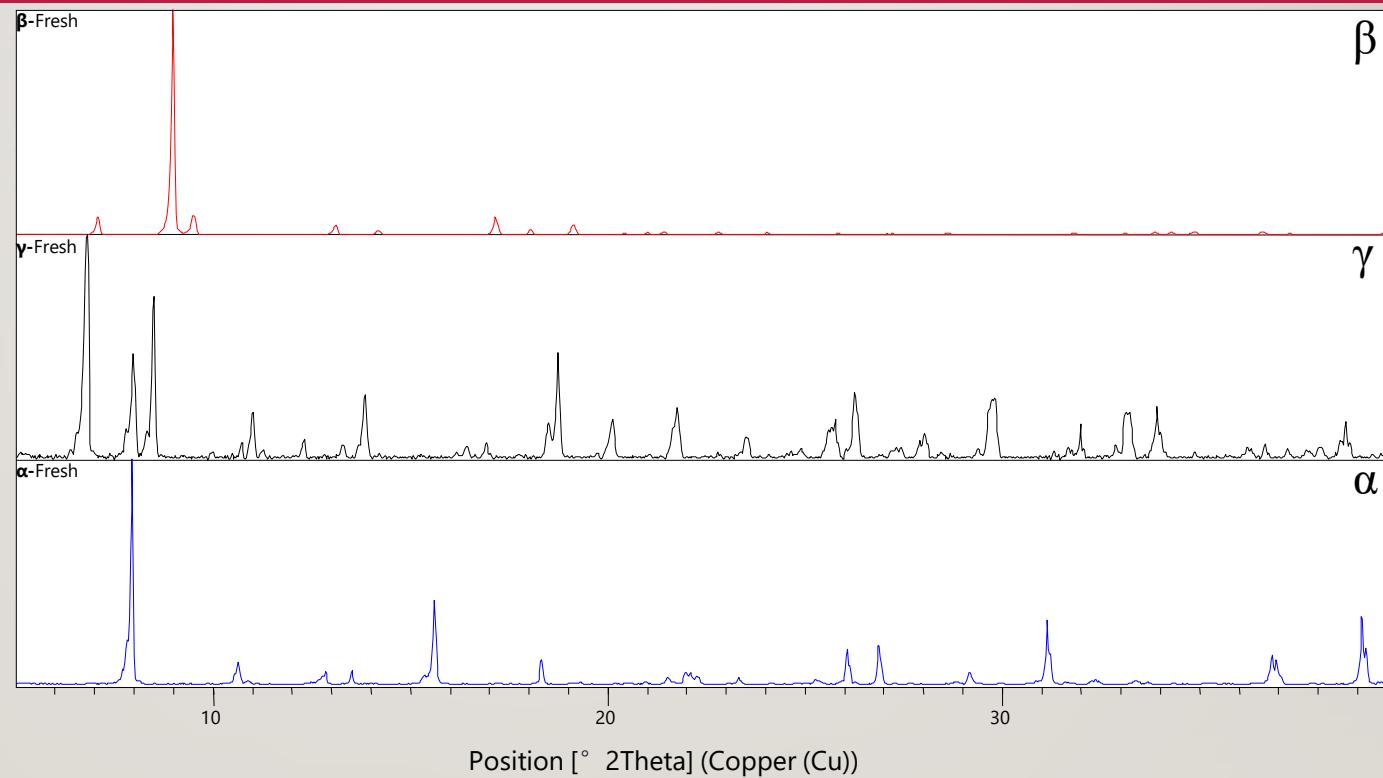
β



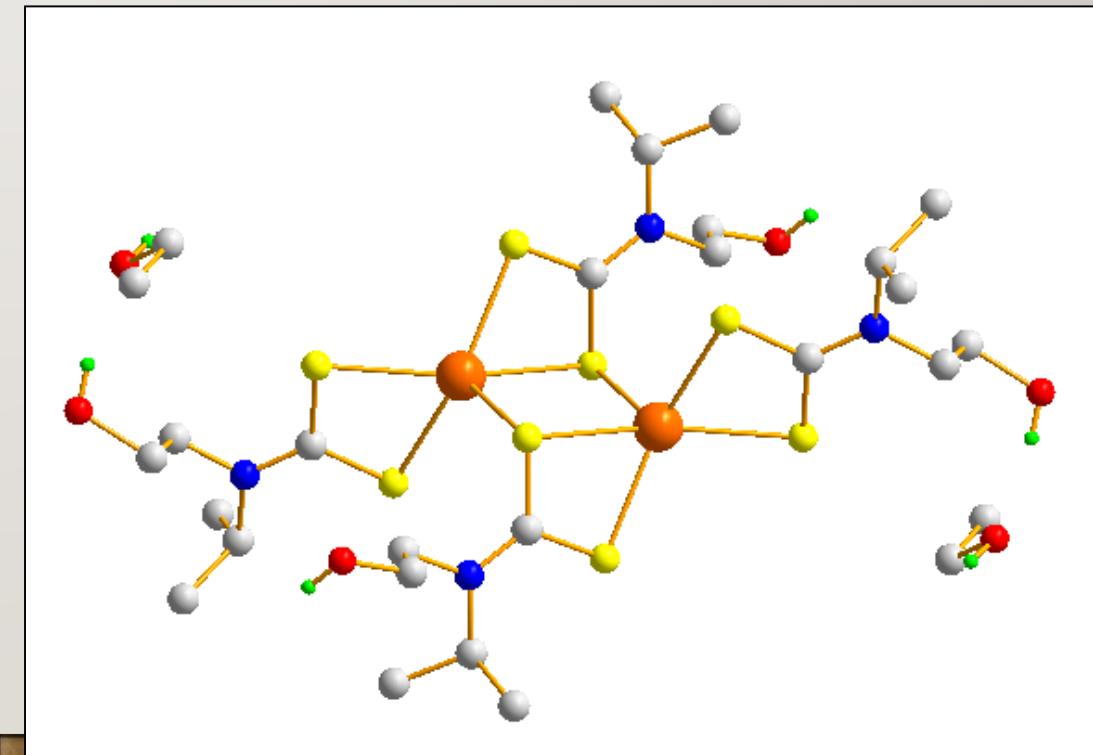
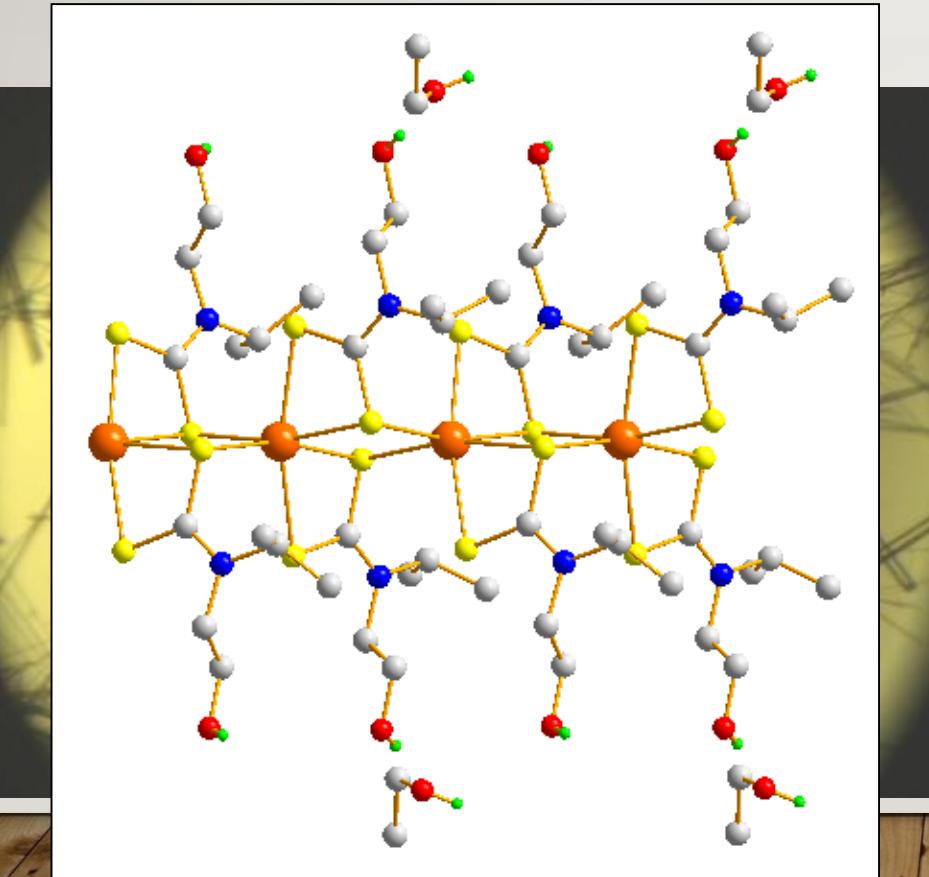
Day6

PHASE CHANGE

- PXRD Pattern



SLOW EVAROPATION (ETHANOL)



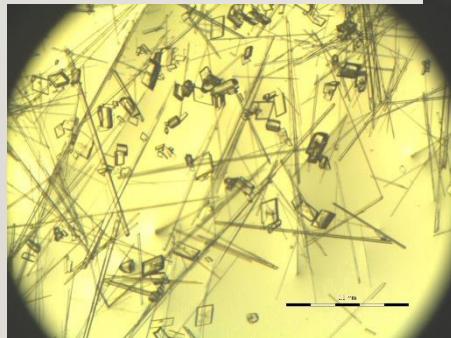
CRYSTAL TRANSFORMATION 2

3rd Hour



δ

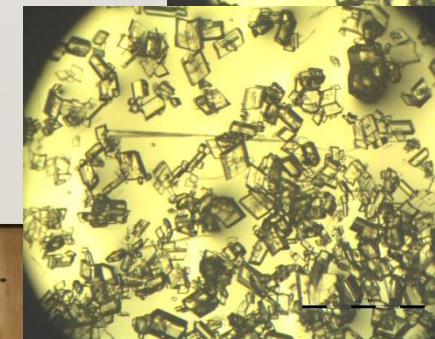
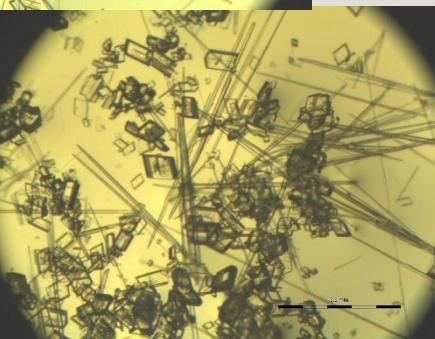
4th Hour



ϵ

Day 3

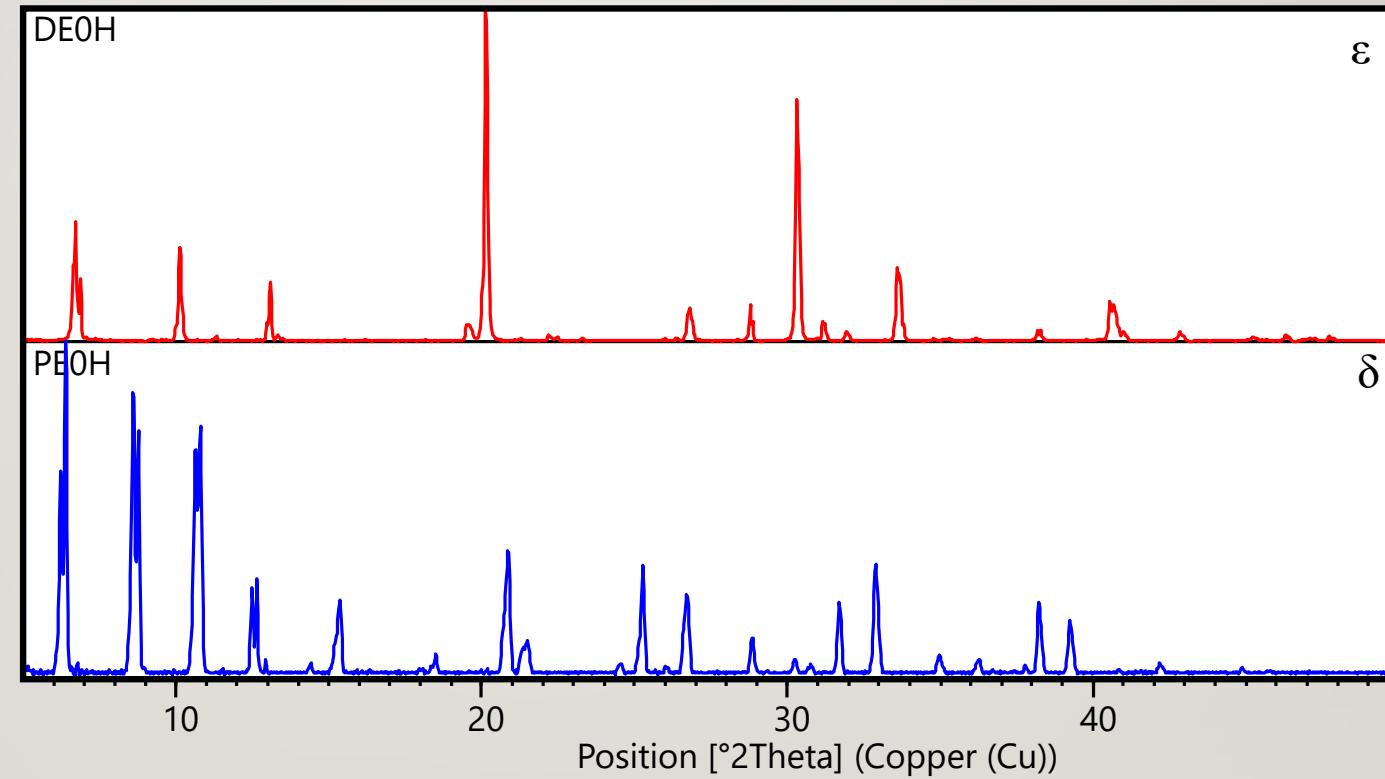
5th HOUR



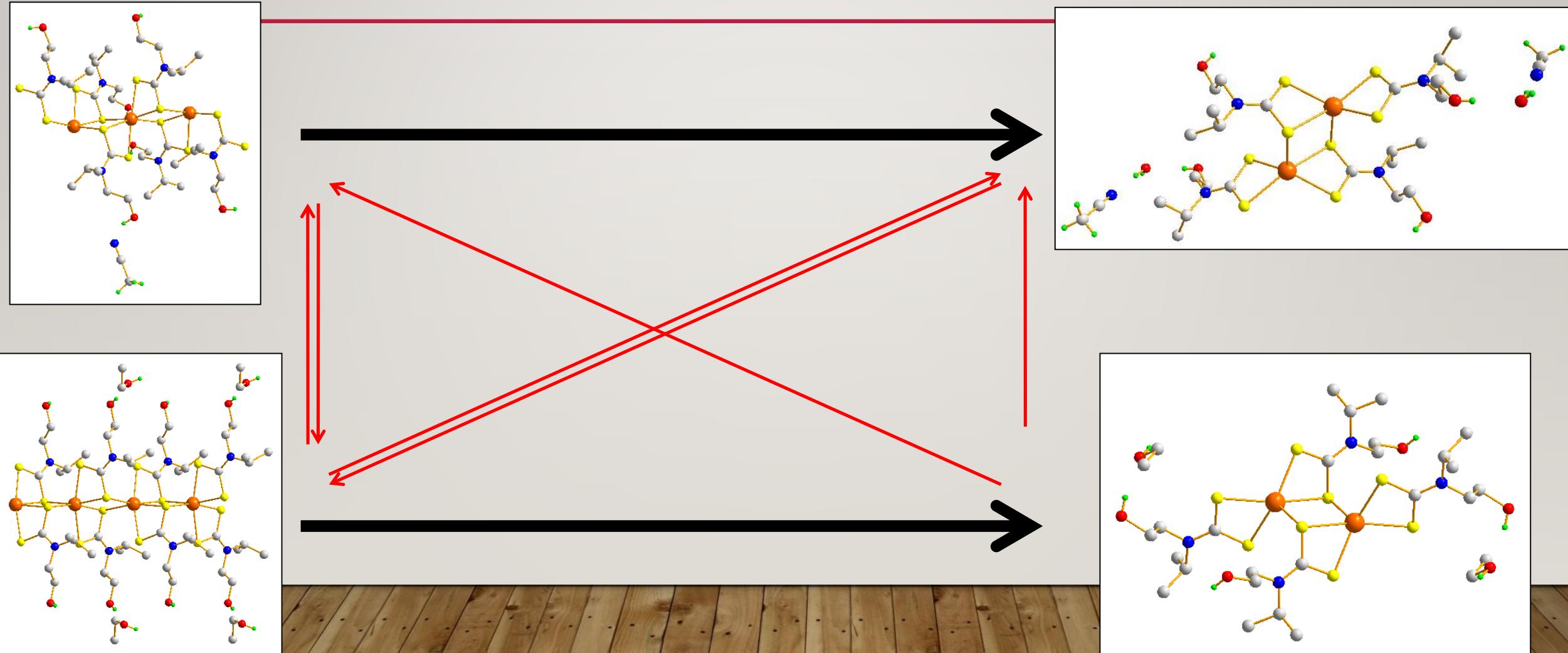
6th Hour

PHASE CHANGE

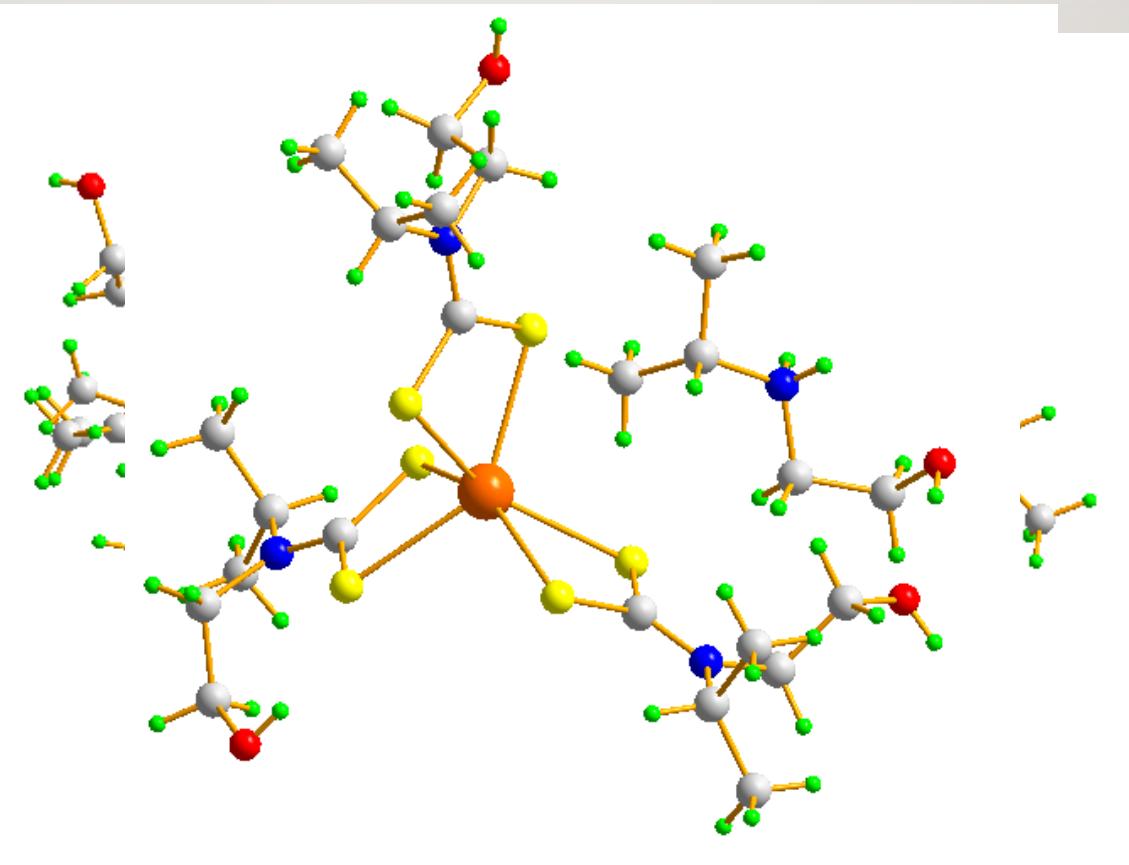
- PXRD Pattern



SOLVENT INDUCE REVERSIBLE TRANSFORMATION



OTHER CRYSTALLISATION OUTCOME



Salt crystal

CONCLUSION

- Fail?
 - Solvate free
- Observation
 - Crystallisation Process
- Various method
 - Polymorph

ACKNOWLEDGEMENT

- Sunway University
- AsCA 2016
- Professor Dr Edward R. T. Tiekkink
- Dr Siti Nadiah Binti Abdul Halim

