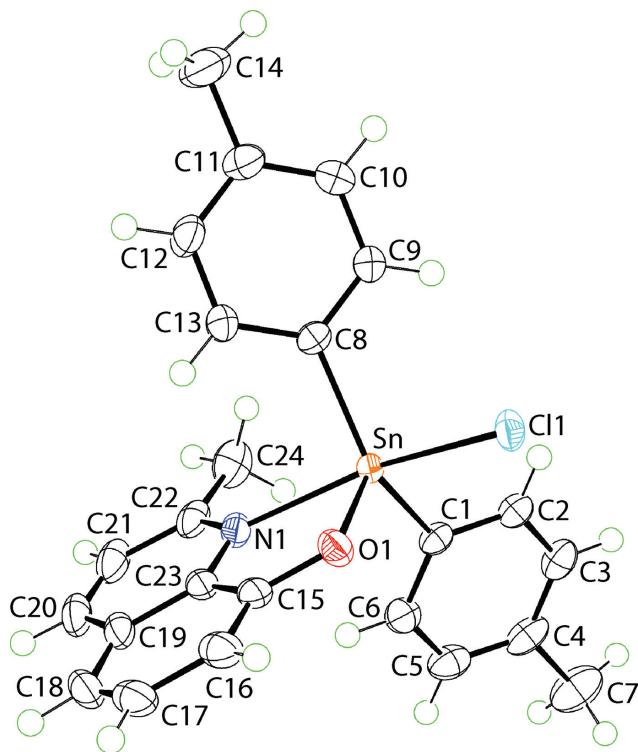


See Mun Lee, Kong Mun Lo and Edward R.T. Tiekink*

Crystal structure of chlorido(2-methylquinolin-8-olato- κ^2N,O)-bis(4-tolyl- κC)tin(IV), $C_{24}H_{22}ClNO_2Sn$



<https://doi.org/10.1515/ncks-2020-0379>

Received July 21, 2020; accepted August 13, 2020; available online August 27, 2020

Abstract

$C_{24}H_{22}ClNO_2Sn$, monoclinic, $P2_1/c$ (no. 14), $a = 13.7584(1)$ Å, $b = 13.5542(1)$ Å, $c = 13.0912(1)$ Å, $\beta = 115.778(1)$ °, $V = 2198.36(3)$ Å³, $Z = 4$, $R_{\text{gt}}(F) = 0.0170$, $wR_{\text{ref}}(F^2) = 0.0448$, $T = 100(2)$ K.

CCDC no.: 2023118

*Corresponding author: Edward R.T. Tiekink, Research Centre for Crystalline Materials, School of Science and Technology, Sunway University, 47500 Bandar Sunway, Selangor Darul Ehsan, Malaysia, e-mail: edwardt@sunway.edu.my. <https://orcid.org/0000-0003-1401-1520>

See Mun Lee and Kong Mun Lo: Research Centre for Crystalline Materials, School of Science and Technology, Sunway University, 47500 Bandar Sunway, Selangor Darul Ehsan, Malaysia

The molecular structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

Table 1: Data collection and handling.

Crystal:	Colourless prism
Size:	0.17 × 0.08 × 0.05 mm
Wavelength:	Cu $K\alpha$ radiation (1.54184 Å)
μ :	10.5 mm ⁻¹
Diffractometer, scan mode:	XtaLAB Synergy, ω
θ_{max} , completeness:	67.1°, >99%
$N(hkl)_{\text{measured}}, N(hkl)_{\text{unique}}, R_{\text{int}}$:	28037, 3934, 0.031
Criterion for $I_{\text{obs}}, N(hkl)_{\text{gt}}$:	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$, 3844
$N(\text{param})_{\text{refined}}$:	256
Programs:	CrysAlis ^{PRO} [1], SHELX [2, 3], WinGX/ORTEP [4]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å²).

Atom	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
Sn	0.72920(2)	0.64037(2)	0.63444(2)	0.01363(5)
O1	0.86273(10)	0.63982(8)	0.60400(10)	0.0197(3)
N1	0.82987(11)	0.77392(10)	0.73685(11)	0.0187(3)
C1	0.75311(13)	0.56365(12)	0.78542(13)	0.0167(3)
C2	0.68405(14)	0.48701(13)	0.78162(14)	0.0190(3)
H2	0.621472	0.473996	0.713299	0.023*
C3	0.70618(15)	0.42933(14)	0.87740(15)	0.0234(4)
H3	0.657938	0.377950	0.874106	0.028*
C4	0.79841(16)	0.44632(14)	0.97795(15)	0.0249(4)
C5	0.86533(15)	0.52438(14)	0.98207(14)	0.0251(4)
H5	0.927191	0.538126	1.050795	0.030*
C6	0.84329(14)	0.58265(13)	0.88735(14)	0.0211(3)
H6	0.889960	0.635862	0.891953	0.025*
C7	0.8280(2)	0.37843(18)	1.07858(19)	0.0394(5)
H7A	0.882347	0.330887	1.080160	0.059*
H7B	0.763560	0.343017	1.072220	0.059*
H7C	0.857471	0.417349	1.148638	0.059*
C8	0.60402(13)	0.73425(13)	0.52452(13)	0.0168(3)
C9	0.50077(14)	0.69628(13)	0.46387(15)	0.0225(4)
H9	0.486346	0.629564	0.474845	0.027*
C10	0.41852(15)	0.75483(14)	0.38742(15)	0.0269(4)
H10	0.348295	0.727760	0.347320	0.032*
C11	0.43723(16)	0.85235(13)	0.36861(16)	0.0239(4)
C12	0.54024(15)	0.89055(14)	0.43056(16)	0.0242(4)

Table 2 (continued)

Atom	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
H12	0.554276	0.957488	0.420034	0.029*
C13	0.62299(14)	0.83262(13)	0.50754(15)	0.0205(3)
H13	0.692821	0.860127	0.548804	0.025*
C14	0.34981(17)	0.91637(17)	0.28329(18)	0.0370(5)
H14A	0.282656	0.878462	0.247722	0.056*
H14B	0.371713	0.938123	0.224987	0.056*
H14C	0.338319	0.974098	0.321785	0.056*
C15	0.93604(13)	0.71141(13)	0.64949(13)	0.0192(3)
C16	1.02651(14)	0.71859(15)	0.63003(15)	0.0247(4)
H16	1.037420	0.672488	0.581341	0.030*
C17	1.10267(15)	0.79356(16)	0.68166(16)	0.0302(4)
H17	1.164393	0.797085	0.666969	0.036*
C18	1.09071(16)	0.86170(14)	0.75250(17)	0.0312(5)
H18	1.144271	0.910796	0.787588	0.037*
C19	0.99817(16)	0.85846(13)	0.77302(15)	0.0253(4)
C20	0.97390(16)	0.92612(14)	0.84113(15)	0.0308(4)
H20	1.022204	0.978787	0.877451	0.037*
C21	0.88156(17)	0.91619(14)	0.85498(15)	0.0290(4)
H21	0.866043	0.962195	0.900629	0.035*
C22	0.80859(15)	0.83784(13)	0.80183(14)	0.0225(4)
C23	0.92151(13)	0.78269(13)	0.72143(13)	0.0195(3)
C24	0.70778(17)	0.82378(15)	0.81613(16)	0.0291(4)
H24A	0.646068	0.818612	0.741396	0.044*
H24B	0.697444	0.880267	0.857099	0.044*
H24C	0.713527	0.763223	0.859287	0.044*
Cl1	0.67328(3)	0.49396(3)	0.51290(3)	0.01955(9)

Source of material

4-Methylphenylmagnesium bromide was prepared from the Grignard reaction of magnesium (Merck) and 4-bromotoluene (Fluka) in tetrahydrofuran. Tetra(4-methylphenyl)tin was synthesised from the reaction of stannic chloride (Fluka) with 4-methylphenylmagnesium bromide in a 1:4 molar ratio. Subsequently, di(4-methylphenyl)tin dichloride was synthesized from the comproportionation reaction of tetra(4-methylphenyl)tin with stannic chloride (Fluka) in a 1:1 molar ratio to obtain a white precipitate. Di(4-methylphenyl)tin dichloride (0.37 g, 1.0 mmol) and 2-methyl 8-quinolinol (Sigma-Aldrich, 0.16 g, 1.0 mmol) were heated in 95% ethanol (30 mL) for 1 h. After filtration, the yellow filtrate was evaporated slowly until colourless crystalline compound was formed. Yield: 0.35 g (70.8%). **M. pt** (Stuart SMP30 digital melting point apparatus; uncorrected): 466–468 K. **IR** (Bruker Vertex 70v FTIR Spectrometer; cm⁻¹): 1505 (m) ν(C=C), 1457 (m) ν(C=N), 1109 (m) ν(C—O). **¹H NMR** (Bruker Ascend 400 MHz NMR spectrometer; CDCl₃; ppm relative to Me₄Si): δ 2.21 (s, 3H, quin-CH₃), 2.30 (s, 6H, CH₃), 7.13–7.59 (m, 12H, Ph-H), 8.35 (d, 1H, Ph-H). **¹³C{¹H} NMR** (as for ¹H NMR): 21.4 (CH₃), 23.5 (quin-CH₃), 115.3, 115.9, 123.7, 127.5, 129.3, 129.7, 130.2, 137.5, 139.3, 140.1, 140.5, 155.5, 156.2 (Ph-C).

Experimental details

The C-bound H atoms were geometrically placed (C—H = 0.95–0.98 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}(\text{C})$.

Comment

8-Hydroxyquinolinato (oxine) compounds of diorganotin chlorides, i.e. molecules of the general formula R₂Sn(oxine)Cl, adopt three distinct structural motifs in their crystals, despite their being only four examples. Two examples, i.e. with R = Et [5] and 2-ClC₆H₄CH₂ [6], feature an N,O-chelated oxine anion and a penta-coordinate *cis*-C₂ClNO donor set. In the case when R = MeOC(=O)CH₂CH₂ [7], i.e. where there is additional coordination potential in the Sn-bound R groups, indeed, one of the carbonyl-O forms a bond with the Sn atom to increase the coordination number to six. The resultant C₂ClNO₂ donor set defines a skew-trapezoidal bipyramidal geometry with the Sn-bound C atoms lying over the Sn—N and Sn—O(R) bonds [7]. Finally, when the R groups are no longer equivalent, i.e. R = Me and R' = Ph, dimerisation occurs as the phenoxide-O atom bridges two Sn atoms [8]. In this case, the resultant C₂ClNO₂ donor set defines a distorted octahedral geometry. There appears to be a sole example of related compound with a 2-methyl substituent in the oxine ligand, namely that of Me(Ph)Sn(Meoxine)Cl [9]. Unlike the aforementioned dimeric structure, this is mononuclear and resembles the penta-coordinated, *cis*-C₂ClNO species above. The present report describes the synthesis and crystal structure determination of a second derivative in the R₂Sn(Meoxine)Cl series, i.e. (4-tolyl)₂Sn(Meoxine)Cl, (I), in continuation of related structural studies [10].

The molecular structure of (I) is shown in the figure (70% displacement ellipsoids) and reveals the Sn atom to be penta-coordinated within a *cis*-C₂ClNO donor set defined by *ipso*-C [Sn—C1 = 2.1285(15) Å and Sn—C8 = 2.1214(16) Å], pyridyl-N [Sn—N1 = 2.3189(14) Å], phenoxide-O [Sn—O1 = 2.0424(12) Å] and Cl [Sn—Cl1 = 2.4489(4) Å] atoms. The coordination geometry is highly distorted with the widest angle of 132.45(6)° corresponding to the C1—Sn—C8 angle and the narrowest angle being the chelate angle, i.e. O1—Sn—N1 = 75.75(5)°. A quantitative assessment of the five-coordinate geometry can be made by calculating the value of τ, which computes to 0.0 for an ideal square-pyramidal geometry and 1.0 for trigonal-bipyramidal [11]. For (I), τ = 0.48, i.e. almost exactly intermediate between the two ideal geometries. Equivalent calculations were performed for Me(Ph)Sn(Meoxine)Cl [9], for which two molecules comprise the asymmetric unit, resulting in τ = 0.38 and 0.56, indicating a tendency to square- and trigonal-bipyramidal geometries, respectively. For completeness, the mononuclear species with oxine, i.e. R = Et [5] and 2-ClC₆H₄ [6] were evaluated, giving

$\tau = 0.36$ and 0.33 , respectively, i.e. indicating distortions towards a square-pyramidal geometry.

In the crystal, the chlorido ligand accepts C—H interactions from a 4-tolyl ring and oxine-methyl [C9—H9 \cdots Cl1ⁱ: H9 \cdots Cl1ⁱ = 2.82 Å, C9 \cdots Cl1ⁱ = 3.621(2) Å with angle at H9 = 142° and C24—H24b \cdots Cl1ⁱⁱ: H24b \cdots Cl1ⁱⁱ = 2.79 Å, C24 \cdots Cl1ⁱⁱ = 3.745(2) Å with angle at H24b = 166° for symmetry operations (i) $1 - x, 1 - y, 1 - z$ and (ii) $x, 3/2 - y, 1/2 + z$] to generate a supramolecular layer in the *bc*-plane. Layers stack along the *a*-axis without directional interactions between them.

A complimentary analysis of the supramolecular aggregation was conducted by calculating the Hirshfeld surface along with the full and decomposed two-dimensional fingerprint plots, with the use of the program Crystal Explorer 17 [12] and guided by literature protocols [13]. The major contribution to the calculated Hirshfeld surface comes from H \cdots H contacts, at 51.7%. This is followed by H \cdots C/C \cdots H contacts at 30.6% with smaller contributions from H \cdots Cl/Cl \cdots H [12.2%] and H \cdots O/O \cdots H [5.3%].

Acknowledgements: Sunway University Sdn Bhd is thanked for financial support of this work through Grant No. STR-RCTR-RCCM-001-2019.

References

- Rigaku Oxford Diffraction: CrysAlis^{PRO}. Rigaku Corporation, Oxford, UK (2018).
- Sheldrick, G. M.: A short history of SHELX. *Acta Crystallogr. A* **64** (2008) 112–122.
- Sheldrick, G. M.: Crystal structure refinement with SHELXL. *Acta Crystallogr. C* **71** (2015) 3–8.
- Farrugia, L. J.: WinGX and ORTEP for Windows: an update. *J. Appl. Cryst.* **45** (2012) 849–854.
- Shi, D.; Hu, S.: Structural studies of chloro(quinoline-8-olato)diethytin(IV), [SnEt₂Cl(C₈H₆NO)]. *Chin. J. Struct. Chem.* **6** (1987) 193–197.
- Keng, T. C.; Lo, K. M.; Ng, S. W.: Chloridobis(2-chlorobenzyl)(quinolin-8-olato- $\kappa^2 N,O$)tin(IV). *Acta Crystallogr. E* **67** (2011) m663.
- Ng, S. W.; Chen Wei, C.; Kumar Das, V. G.; Charland, J.-P.; Smith, F. E.: Six-coordinate tin in a dialkylchlorooxatinotin(IV) complex: skew-trapezoidal bipyramidal bis(2-carbomethoxyethyl)chloro(quinolin-8-olato)tin(IV). *J. Organomet. Chem.* **364** (1989) 343–351.
- Vafaee, M., Amini, M. M.; Ng, S. W.: Bis(μ -quinolin-8-olato)- $\kappa^3 N,O;O;\kappa^3 O:N,O$ -bis[chloridomethylphenyltin(IV)]. *Acta Crystallogr. E* **66** (2010) m964.
- Vafaee, M., Amini, M. M.; Ng, S. W.: Chloridomethyl(2-methylquinolin-8-olato- $\kappa^2 N,O$)phenyltin(IV). *Acta Crystallogr. E* **66** (2010) m965.
- Lee, S.; Lo, K.; Tiekkink, E. R. T.: Crystal structure of chlorido-4-fluorobenzyl-bis(2-methylquinolin-8-olato- $\kappa^2 N,O$)tin(IV), C₂₇H₂₂ClF₂N₂O₂Sn, Z. *Kristallogr. NCS* **234** (2019) 823–825.
- Addison, A. W.; Rao, T. N.; Reedijk, J.; van Rijn, J.; Verschoor, G. C.: Synthesis, structure, and spectroscopic properties of copper(II) compounds containing nitrogen-sulphur donor ligands; the crystal and molecular structure of aqua[1,7-bis(*N*-methylbenzimidazol-2'-yl)-2,6-dithiaheptane]-copper(II) perchlorate. *J. Chem. Soc., Dalton Trans.* (1984) 1349–1356.
- Turner, M. J.; McKinnon, J. J.; Wolff, S. K.; Grimwood, D. J.; Spackman, P. R.; Jayatilaka, D.; Spackman, M. A.: Crystal Explorer v17. The University of Western Australia, Australia (2017).
- Tan, S. L.; Jotani, M. M.; Tiekkink, E. R. T.: Utilizing Hirshfeld surface calculations, non-covalent interaction (NCI) plots and the calculation of interaction energies in the analysis of molecular packing. *Acta Crystallogr. E* **75** (2019) 308–318.