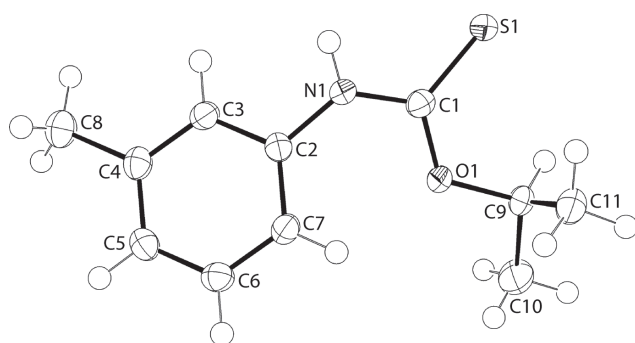


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# Crystal structure of *N*-(3-methylphenyl)(propan-2-yloxy)carbothioamide, C<sub>11</sub>H<sub>15</sub>NOS



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## Abstract

C<sub>11</sub>H<sub>15</sub>NOS, monoclinic, *C*2/*c* (no. 15),  $a = 21.0452(12)$  Å,  $b = 6.1447(4)$  Å,  $c = 17.1892(10)$  Å,  $\beta = 93.3520(10)^\circ$ ,  $V = 2219.0(2)$  Å<sup>3</sup>,  $Z = 8$ ,  $R_{\text{gt}}(F) = 0.0291$ ,  $wR_{\text{ref}}(F^2) = 0.0854$ ,  $T = 100(2)$  K.

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The molecular structure is shown in the figure. Table 1 contains crystallographic data and Table 2 contains the list of the atoms including atomic coordinates and displacement parameters.

## Source of material

All chemicals and solvents were used as purchased without purification. The reaction was carried out under ambient conditions. The melting point was determined on a Krüss KSP1N melting point meter. The IR spectrum was obtained on a Perkin Elmer Spectrum 400 FT Mid-IR/Far-IR spectrophotometer from 4000 to 400 cm<sup>-1</sup>; abbreviation: s, strong.

Preparation of the title compound: *m*-Tolyl isothiocyanate (Merck; 2.5 mmol, 0.34 mL) was added to NaOH

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Table 1: Data collection and handling.

Crystal:	Colourless prism
Size:	0.20 × 0.11 × 0.09 mm
Wavelength:	Mo K $\alpha$ radiation (0.71073 Å)
$\mu$ :	0.26 mm <sup>-1</sup>
Diffractometer, scan mode:	Bruker SMART APEX, $\omega$
$\theta_{\text{max}}$ , completeness:	27.5°, >99%
$N(hkl)_{\text{measured}}$ , $N(hkl)_{\text{unique}}$ , $R_{\text{int}}$ :	13386, 2545, 0.024
Criterion for $I_{\text{obs}}$ , $N(hkl)_{\text{gt}}$ :	$I_{\text{obs}} > 2 \sigma(I_{\text{obs}})$ , 2283
$N(\text{param})_{\text{refined}}$ :	133
Programs:	Bruker [1], SHELX [2–4], WinGX/ORTEP [5]

Table 2: Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å<sup>2</sup>).

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.35255(2)	−0.21699(5)	0.00951(2)	0.01905(10)
O1	0.36789(4)	0.15300(15)	0.08977(5)	0.01890(19)
N1	0.26773(5)	0.03575(17)	0.07219(6)	0.0165(2)
H1N	0.2418(6)	−0.063(2)	0.0530(8)	0.020*
C1	0.32909(5)	0.00139(19)	0.05886(6)	0.0157(2)
C2	0.23881(5)	0.19710(19)	0.11801(6)	0.0148(2)
C3	0.18019(5)	0.1410(2)	0.14645(7)	0.0171(2)
H3	0.1625	0.0017	0.1348	0.021*
C4	0.14727(6)	0.2859(2)	0.19151(7)	0.0186(2)
C5	0.17389(6)	0.4890(2)	0.20825(7)	0.0188(2)
H5	0.1523	0.5892	0.2395	0.023*
C6	0.23190(6)	0.5457(2)	0.17949(7)	0.0188(2)
H6	0.2495	0.6852	0.1911	0.023*
C7	0.26468(6)	0.40209(19)	0.13412(7)	0.0170(2)
H7	0.3042	0.4429	0.1143	0.020*
C8	0.08445(7)	0.2229(2)	0.22276(9)	0.0296(3)
H8A	0.0848	0.2582	0.2784	0.044*
H8B	0.0776	0.0663	0.2156	0.044*
H8C	0.0500	0.3032	0.1946	0.044*
C9	0.43606(5)	0.1434(2)	0.07709(7)	0.0195(3)
H9	0.4510	−0.0111	0.0796	0.023*
C10	0.46806(7)	0.2724(3)	0.14296(9)	0.0334(3)
H10A	0.4568	0.2105	0.1929	0.050*
H10B	0.4538	0.4241	0.1395	0.050*
H10C	0.5143	0.2665	0.1393	0.050*
C11	0.44799(6)	0.2392(2)	−0.00183(8)	0.0236(3)
H11A	0.4280	0.1470	−0.0428	0.035*
H11B	0.4939	0.2468	−0.0081	0.035*
H11C	0.4298	0.3858	−0.0057	0.035*

(Merck; 2.5 mmol, 0.10 g) in *i*-PrOH (Merck; 5 mL) and the mixture was left for stirring at room temperature for 2 h, followed by the addition of excess 5 M HCl solution. The resulting mixture was stirred for a further 1.5 h. The final product was extracted with chloroform (Merck; 20 mL) and left for evaporation at room temperature, yielding brown crystals after 3 weeks. *M.pt.*: 323–325 K. IR (cm<sup>-1</sup>): 3220 (s) ν(N–H), 1491 (s) ν(C–N), 1207 (s) ν(C–O), 1091 (s) ν(C=S).

### Experimental details

The C-bound H atoms were geometrically placed (C–H = 0.95–1.00 Å) and refined as riding with  $U_{\text{iso}}(\text{H}) = 1.2\text{--}1.5U_{\text{eq}}(\text{C})$ . The N-bound H-atom was located in a difference Fourier map but was refined with a distance restraint of N–H = 0.88 ± 0.01 Å, and with  $U_{\text{iso}}(\text{H})$  set to  $1.2U_{\text{equiv}}(\text{N})$ . Owing to poor agreement, the ( $\bar{9}$  1 1) reflection was omitted from the final cycles of refinement.

### Comment

A recent overview of the known crystal structures for molecules of the general formula, ROC(=S)N(H)R'' (R/R'' = alkyl and/or aryl), *i.e.* the alkoxy-carbothioamides, showed a high degree of concordance in their structural features [6]. Most notable was the universal adoption of a thioamide tautomer. The majority of structures featured a *syn*-disposition of the thione-S and thioamide-N–H atoms. The exceptions occur when an anti-disposition of the thione-S and thioamide-N–H atoms is found as, for example, in the structure where R = Me and R'' = 4-C(=O)Me-phenyl [7], allowing for the formation of intermolecular N–H...O hydrogen bonding, and where R = 4-pyridylphenyl and R'' = phenyl [8], allowing for intermolecular N–H...N hydrogen bonding. Herein, as a continuation of structural studies of this class of compound [6, 7, 9], the crystal and molecular structures of the compound with R = <sup>i</sup>Pr and R'' = 3-Me-phenyl are described.

The molecular structure is shown in the figure (70% displacement ellipsoids) and features the normally observed [6] *syn*-disposition of the thione-S and thioamide-N–H atoms. In accord with expectation, the central C1, N1, O1, S1 residue is planar with the r.m.s. deviation being 0.0044 Å. The dihedral angle between the central plane and appended 3-tolyl group is 23.06(5)°. Evidence for the thioamide tautomer is found in the magnitude of the C1=S1 [1.6772(12) Å] and C1–N1 [1.3412(15) Å] bond lengths, and in the location of the thioamide-N–H atom in the crystallographic refinement. The angles subtended at the quaternary-C1 atom follow the expected trends with S1–C1–O1 [124.81(9)°] being wider than S1–C1–N1 [121.82(9)°] and each of these being wider than O1–C1–N1 [113.36(10)°].

The *syn*-disposition of the thione-S1 and thioamide-N–H atoms allows for the formation of thioamide-N–H...S(thione) hydrogen bonds in the molecular packing [N1–H1n...S1<sup>i</sup>: H1n...S1<sup>i</sup> = 2.584(13) Å, N1...S1<sup>i</sup> = 3.4352(11) Å with angle at H1n = 167.0(11)° for symmetry operation *i*: 1/2 – *x*, –1/2 – *y*, –*z*]. As these occur between centrosymmetrically related molecules, eight-membered {...HNCS}<sub>2</sub> synthons ensue. Intermolecular points of contact of the type tolyl-C–H...π [C6–H6...Cg(C2–C7)<sup>ii</sup>: H6...Cg(C2–C7)<sup>ii</sup> = 2.80 Å, C6...Cg(C2–C7)<sup>ii</sup> = 3.4597(13) Å and angle at H6 = 127° for *ii*: 1/2 – *x*, 1/2 + *y*, 1/2 – *z*] and parallel C=S...π [C1–S1...Cg(C2–C7)<sup>iii</sup>: S1...Cg(C2–C7)<sup>iii</sup> = 3.8929(6) Å, C1...Cg(C2–C7)<sup>iii</sup> = 3.9582(12) Å and angle at S1 = 79.85(4)° for *iii*: 1/2 – *x*, 1/2 – *y*, –*z*], each involving the 3-tolyl ring, serve to link molecules into a supramolecular layer in the *bc*-plane. Layers stack along the *a* axis direction without directional interactions between them.

The most closely related structures in the literature are the polymorphic structures EtOC(=S)N(H)(3-Me-phenyl) [6, 10], *i.e.* where <sup>i</sup>Pr is substituted with Et; both polymorphs are monoclinic. In the *P*<sub>2</sub><sub>1</sub>/*c* form [6], there are two independent molecules in the asymmetric unit whereas in the *C*<sub>2</sub>/*c* form [10], one molecule comprises the asymmetric unit. Not unexpectedly, similar trends in conformation and geometric parameters are evident, along with supramolecular association *via* {...HNCS}<sub>2</sub> synthons, as described above.

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### References

1. Bruker. APEX2 and SAINT. Bruker AXS Inc., Madison, WI, USA (2008).
2. Sheldrick, G. M.: SADABS. University of Göttingen, Germany (1996).
3. Sheldrick, G. M.: A short history of SHELX. *Acta Crystallogr. A* **64** (2008) 112–122.
4. Sheldrick, G. M.: Crystal structure refinement with SHELXL. *Acta Crystallogr. C* **71** (2015) 3–8.
5. Farrugia, L. J.: WinGX and ORTEP for Windows: an update. *J. Appl. Crystallogr.* **45** (2012) 849–854.
6. Jotani, M. M.; Yeo, C. I.; Tiekink, E. R. T.: A new monoclinic polymorph of *N*-(3-methylphenyl) ethoxycarbothioamide: crystal structure and Hirshfeld surface analysis. *Acta Crystallogr. E* **73** (2017) 1889–1897.
7. Ho, S. Y.; Bettens, R. P. A.; Dakternieks, D.; Duthie, A.; Tiekink, E. R. T.: Prevalence of the thioamide {...H–N–C=S}<sub>2</sub> synthon solid-state (X-ray crystallography),

- solution (NMR) and gas-phase (theoretical) structures of *O*-methyl-*N*-aryl-thiocarbamides. *CrystEngComm* **7** (2005) 682–689.
8. Xiao, H.-L.; Wang, K.-F.; Jian, F.-F.: (4-Pyridyl)methyl *N*-phenylthiocarbamate. *Acta Crystallogr.* **E62** (2006) o2852–o2853.
  9. Yeo, C. I.; Tiekink, E. R. T.: Crystal structure of *N*-(2-methylphenyl)ethoxycarbothioamide, C<sub>10</sub>H<sub>13</sub>NOS. *Z. Kristallogr. NCS* **233** (2018) 299–301.
  10. Tadbuppa, P.; Tiekink, E. R. T.: Crystal structure of *o*-ethyl *N*-(*m*-tolyl)thiocarbamate, SC(OC<sub>2</sub>H<sub>5</sub>)NH(C<sub>6</sub>H<sub>4</sub>CH<sub>3</sub>). *Z. Kristallogr. NCS* **220** (2005) 395–396.